

## Review topic: Atomic structure

	Ways to practice skills	R	A	G	Comment
2.2 Atomic structure and the Periodic Table					
1	Describe the structure of the atom as a central nucleus containing neutrons and protons surrounded by electrons in shells	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	
2	State the relative charges and relative masses of a proton, a neutron and an electron	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	
3	Define proton number / atomic number as the number of protons in the nucleus of an atom	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	
4	Define mass number / nucleon number as the total number of protons and neutrons in the nucleus of an atom	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	
5	Determine the electronic configuration of elements with proton number 1 to 20, e.g. 2,8,3	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	